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# Iron chemical speciation in atmospherically processed industrial aerosols

Pascal Flament\*<sup>†1</sup>, Héloïse Delpouve<sup>1</sup>, Karine Deboudt<sup>1</sup>, and Eugène Bychkov<sup>1</sup>

<sup>1</sup>Laboratoire de Physico-Chimie de l'Atmosphère (LPCA) – Université Littoral Côte d'Opale – Bâtiment MREI 2 189A Avenue Maurice Schumann F-59140 Dunkerque, France

## Abstract

How human activities impact the transport of bioavailable iron to the oceans is a fundamental question addressed in atmospheric and oceanographic studies. Fe becomes more soluble during atmospheric transport and when associated with anthropogenic aerosols. However, Fe speciation in these aerosols is poorly known. The Fe solubility strongly depending on its oxidation state, this study focuses on the impact of cloud processes on the Fe chemical speciation for industrial aerosols.

The studied sample was collected on the industrial filter located in the chimney stack of a metallurgy plant (cooling area of the sintering unit). This sample was fully characterized by X-Ray diffraction, laser diffraction particle sizing and scanning electron microscopy (SEM-EDX). Weathering processes are simulated using  $300 \pm 1$  mg of particles into 1L of a leaching solution ( $80 \mu\text{M}$  oxalic acid solution) mimicking cloud droplets. This represents  $16.2 \pm 0.1$  mg of total Fe. The leaching experiment is performed for 90 minutes under simulated solar irradiation. Fe solid state chemical speciation is investigated by Mössbauer spectroscopy before and after the leaching experiment.

In unprocessed particles, the most intense quadrupole doublet (47.7% of the total resonance pattern) of the  $^{57}\text{Fe}$  Mössbauer spectrum corresponds to high-spin  $\text{Fe}^{3+}$  species in oxide environment.  $\alpha\text{-Fe}_2\text{O}_3$  rather well crystallized, with a slight distortion of the Fe local environment, in rhombohedral  $\alpha\text{-Fe}_2\text{O}_3$  lattice and ill-crystallized and impure magnetite  $\text{Fe}_3\text{O}_4$ , are the two other Fe species evidenced by Mössbauer spectroscopy. In the magnetite octahedral sites, Fe sites are partially occupied by metallic cations of similar size ( $\text{Mn}^{2+}$  or  $\text{Ca}^{2+}$ ).

After simulated cloud weathering (about 0.2% of solubilized Fe), only small changes are evidenced in the  $\alpha\text{-Fe}_2\text{O}_3$  and  $\text{Fe}_3\text{O}_4$   $^{57}\text{Fe}$  Mössbauer spectrum hyperfine interaction parameters. This suggests that, if cloud processes can significantly affect the iron solubility, speciation in metallurgy aerosols remains unaffected by medium or long range transport atmospheric processes.

**Keywords:** aerosols, metallurgy, iron speciation, cloud chemistry, simulation, Mössbauer spectroscopy

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\*Speaker

<sup>†</sup>Corresponding author: pascal.flament@univ-littoral.fr